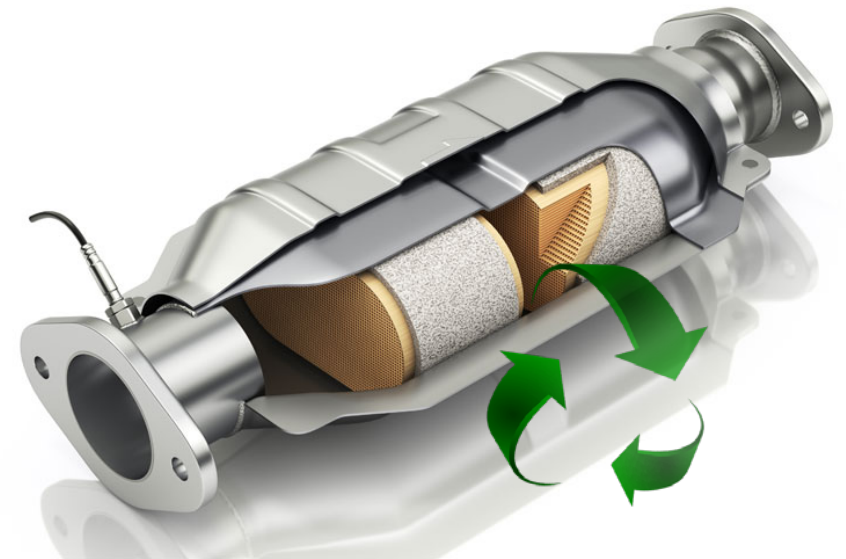


Sustainability and recovery of precious metals from scrapped DPF

Associate professor: Shuang Ma Andersen

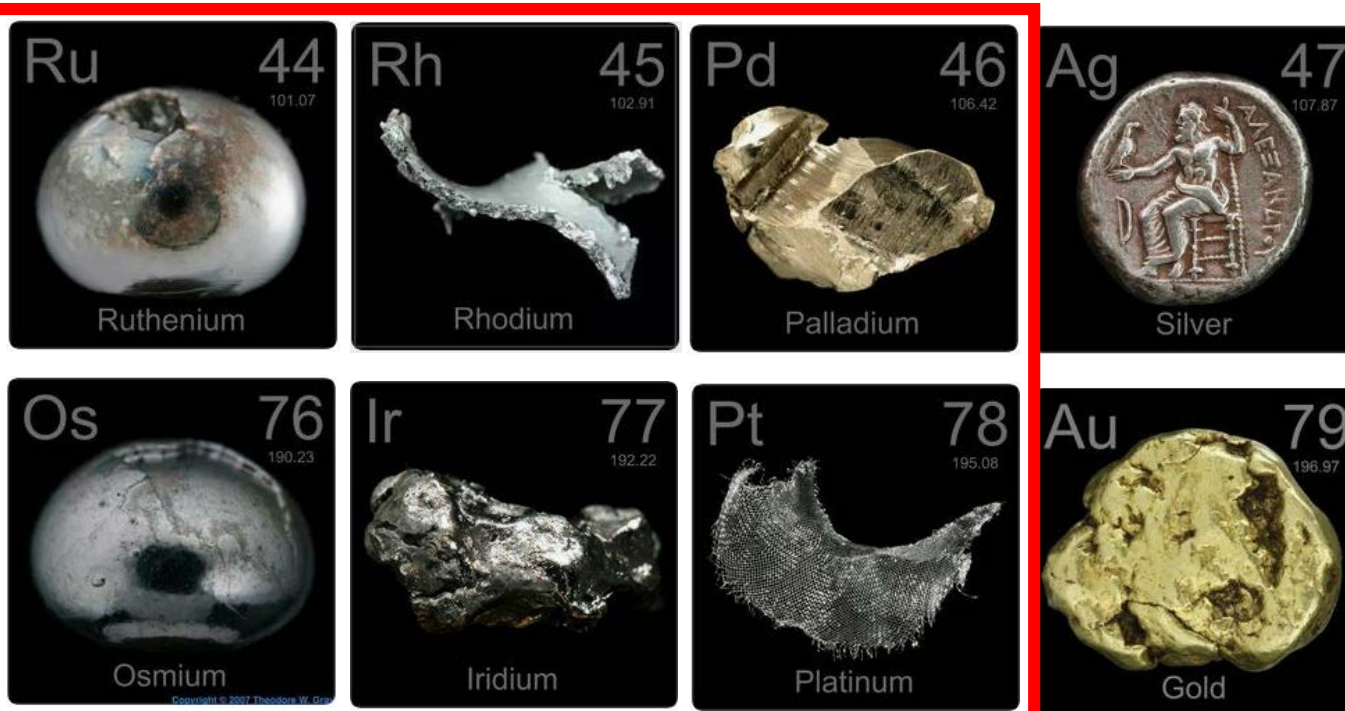
Section: Chemical Engineering
Department of Chemical engineering, Biotechnology & Environmental Technology
University of Southern Denmark
mashu@kbm.dk



Platinum Group Metals (PGMs)

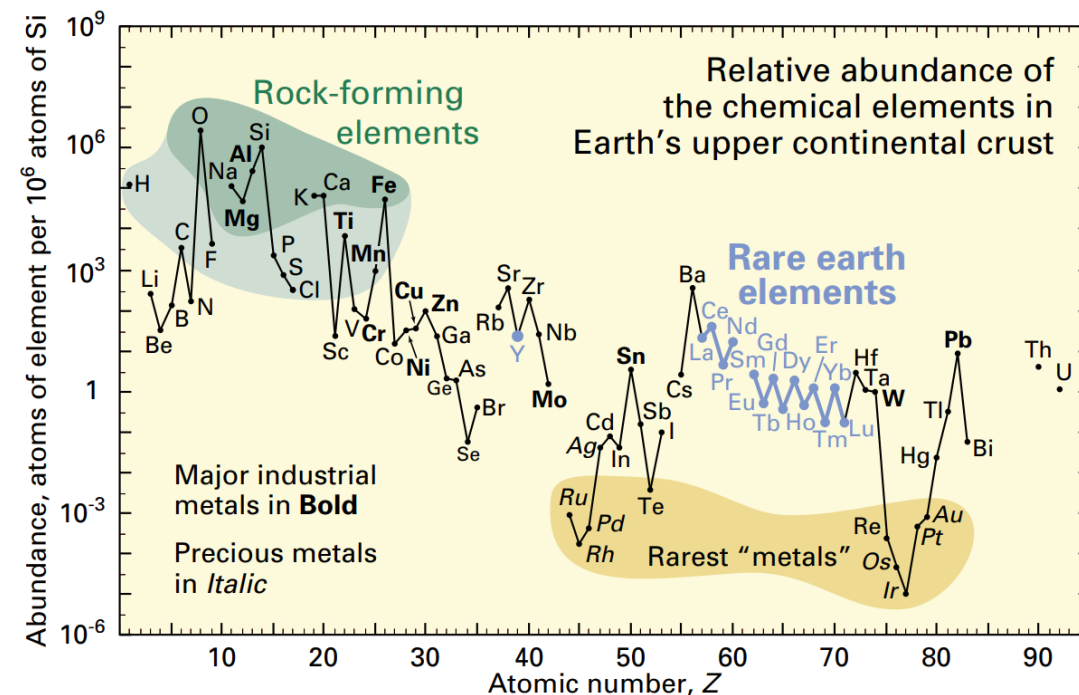
One of the 27 **critical raw materials** identified by EU

One quarter of all goods manufactured either contain a PGM or had a PGM play a key role in its production.

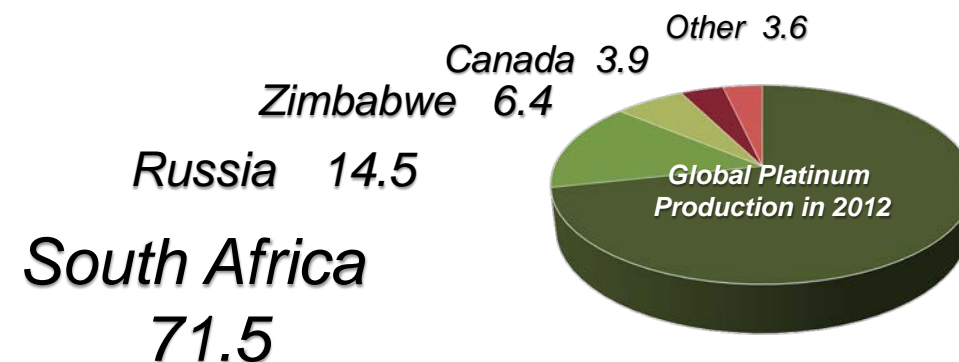


Platinum Group Metals (PGMs)

Precious Metals (PMs)



Abundance (atom fraction) of the chemical elements in Earth's upper continental crust as a function of atomic number.



Gordon B. Haxel, Sara Boore, and Susan Mayfield from USGS

Background

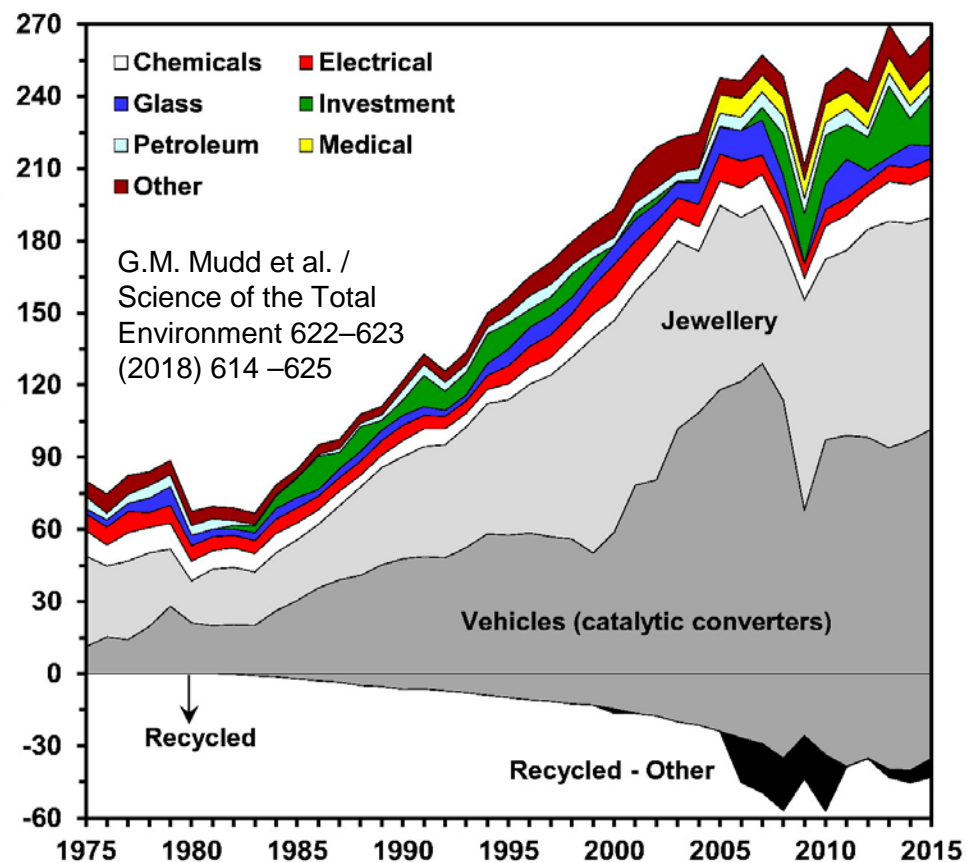
Classic

Mechanisms

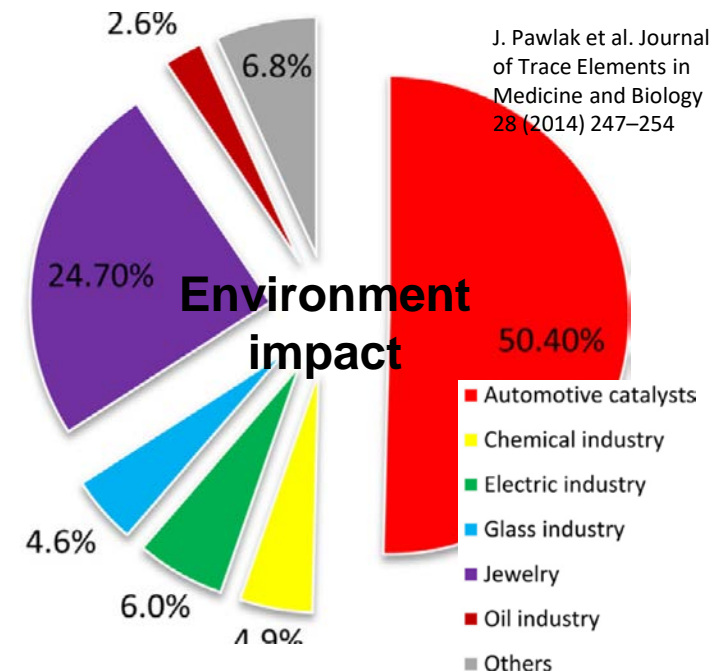
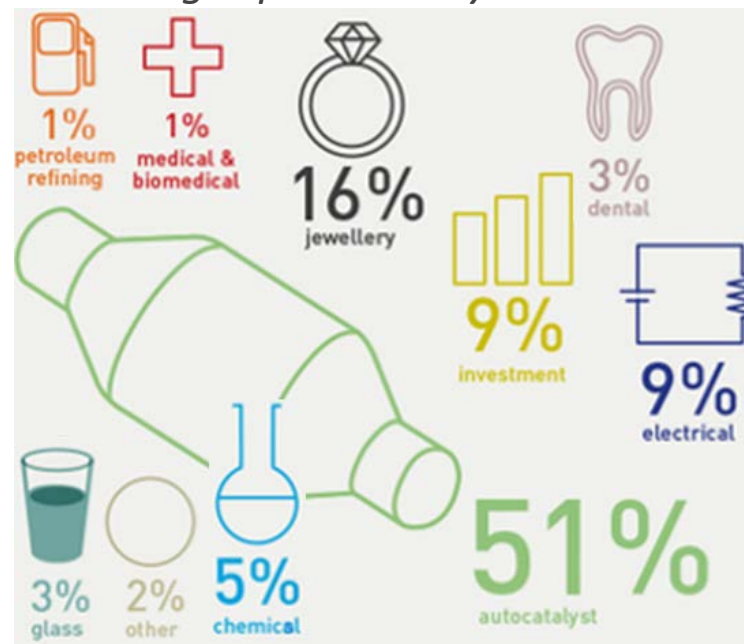
Circular economy

Messages

Platinum Demand by Sector (t Pt/year)



PGM usages per industry as in 2010



Type of plant material	Concentration [ng/g]		
	Pt	Pd	Rh
Grass	8.98	3.20	0.68
Pine, birch	12.0	2.0	2.0
Tomato (without skin)	0.10	–	–
Carrot	0.31	–	–
Cabbage	1.1	–	–
Lettuce	2.1	–	–
Celery	1.3	–	–
Onion (without skin)	0.03	–	–
Dandelion	5.4–30	0.83–1.5	2.0–7.0
Greater plantain	3.6–10.1	0.45–2.1	1.1–3.4
Lichen	30.0	2.4	5.4
Rye grass	4.6–5.8	0.1	2.1–2.2

	Concentration [ng/g]		
	Pt	Pd	Rh
Shellfish tissue (<i>Asellus aquaticus</i>)	0.04–12.4	–	–
Mussel tissue (<i>Dreissena polymorpha</i>)	0.1–0.5	1.0	–
Eel liver (<i>Anguilla anguilla</i>)	–	0.18	–
Fish tissue (<i>Barbus barbus</i>)	0.1–0.4	0.3–7.0	0.1–2.0
Peregrine falcon (<i>Falco peregrinus</i>)	0.5	1.4	0.3
Feathers	2.7	0.8	0.6
Blood	0.4	0.5	0.3
Eggs	0.2	–	–
Feces	0.2	0.7	0.3
Liver	0.2	0.3	0.3
Kidneys	0.2	0.3	0.3



Primary ore: ~E-3 wt%



Auto catalyst: ~E-1wt%

PGM content in selected samples of plants and animal tissues

Current industrial PGM recycling technologies

Pyrometallurgy

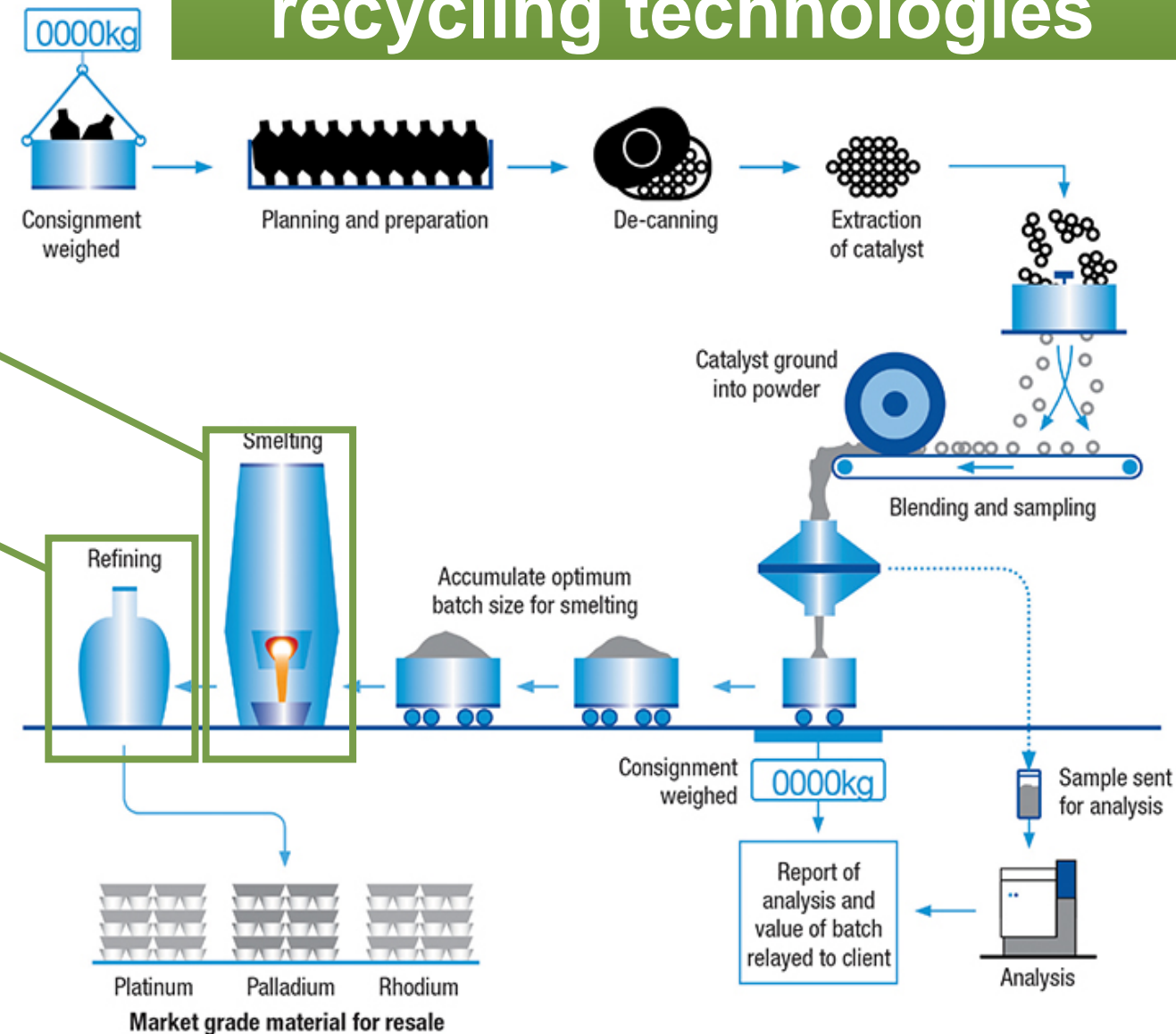


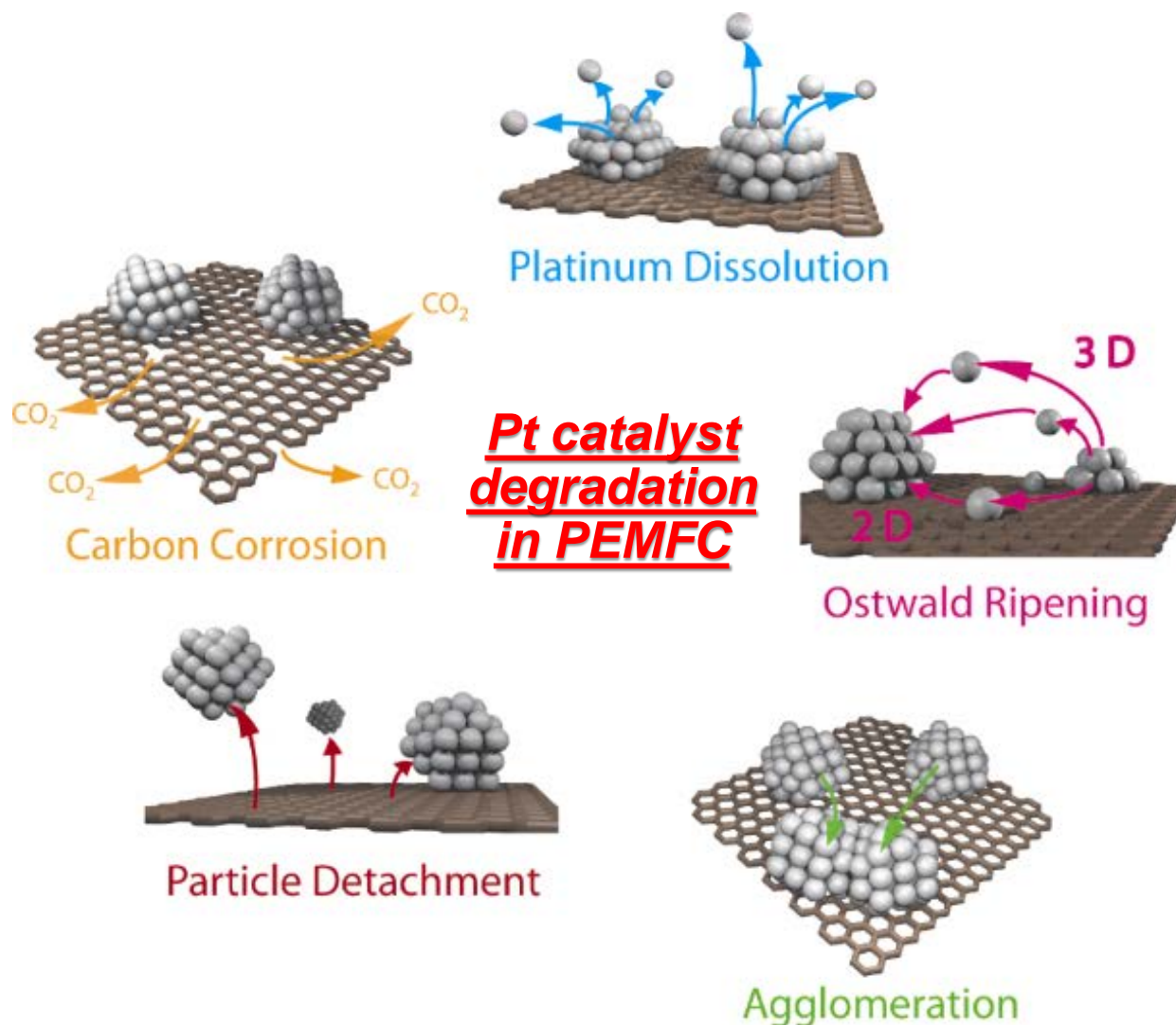
High temperature - energy demanding and may generate harmful substances

Hydrometallurgy

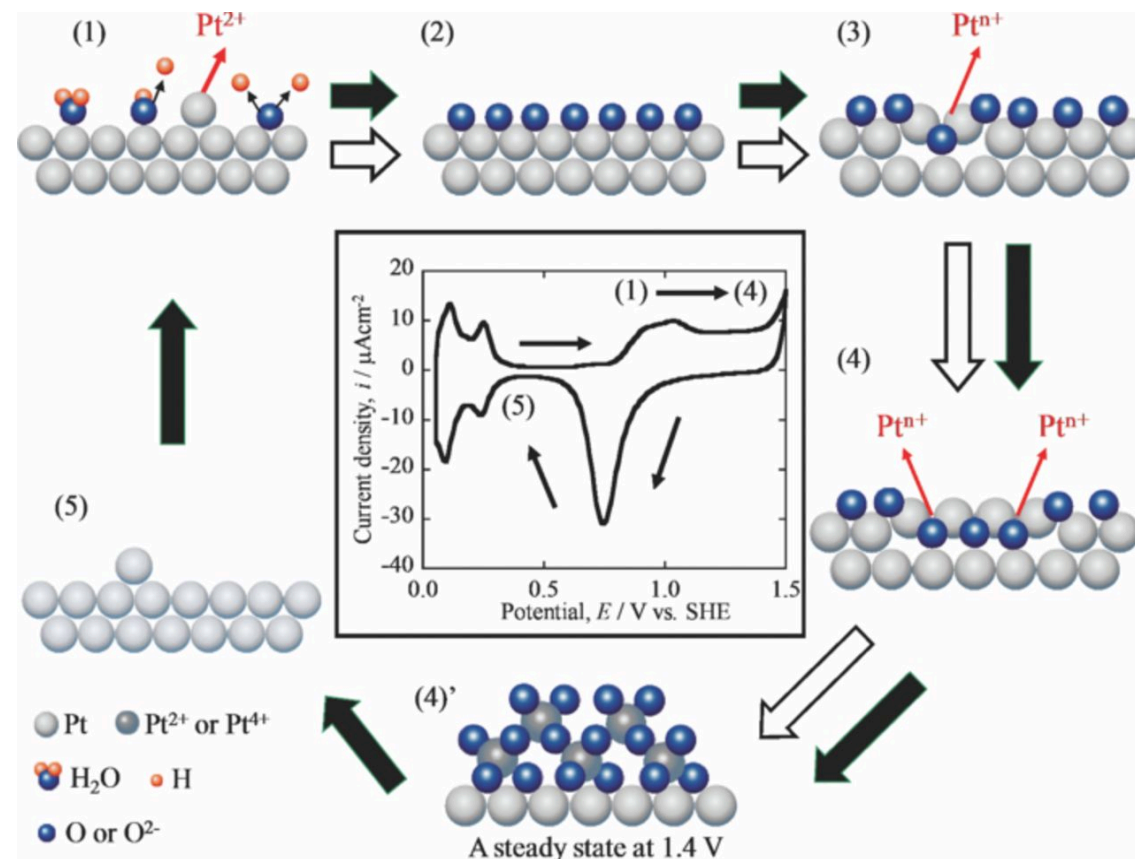


Hazardous chemicals such as the toxic complexing agent NaCN or the corrosive *aqua regia*



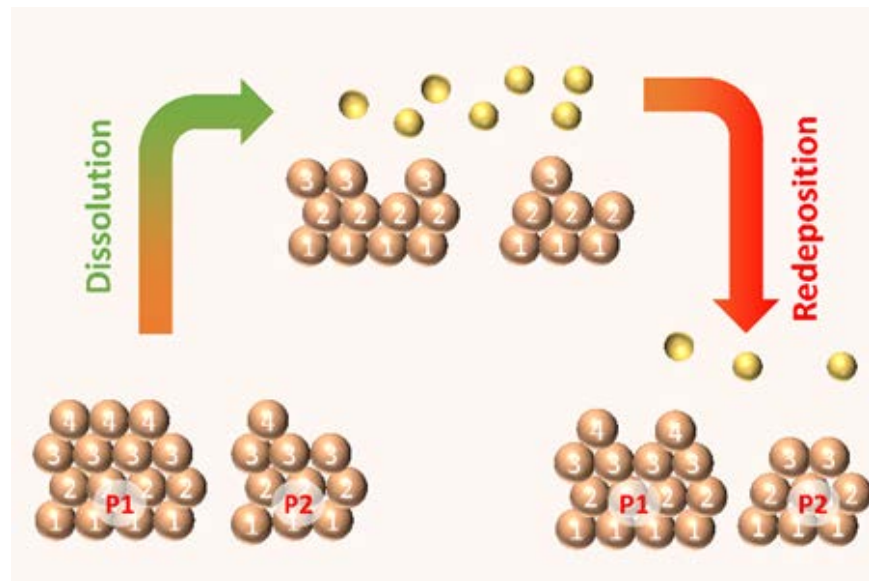


Fundamental principle of platinum dissolution is described by surface reaction dominated by transit potential treatment.



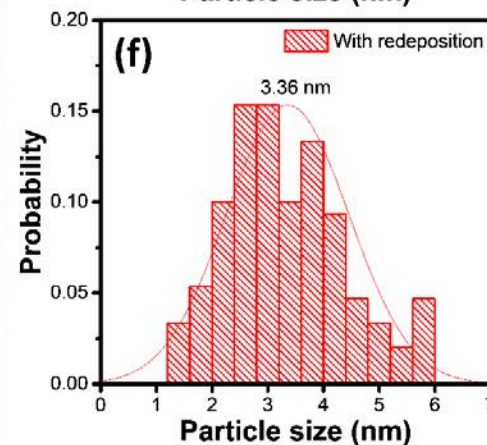
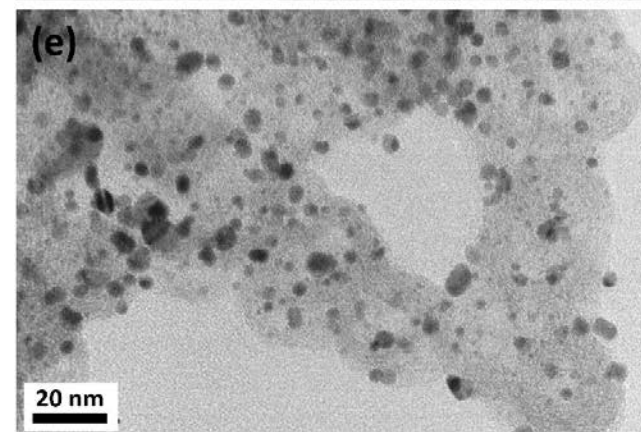
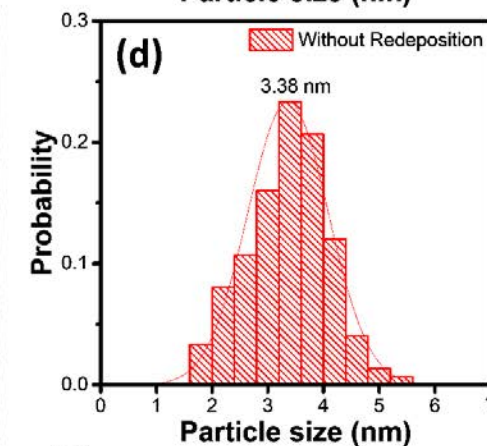
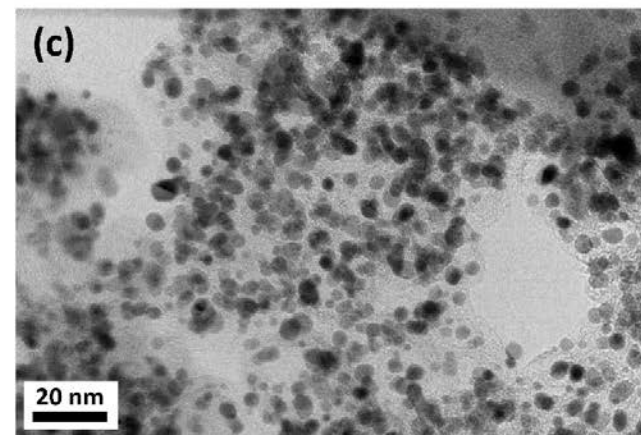
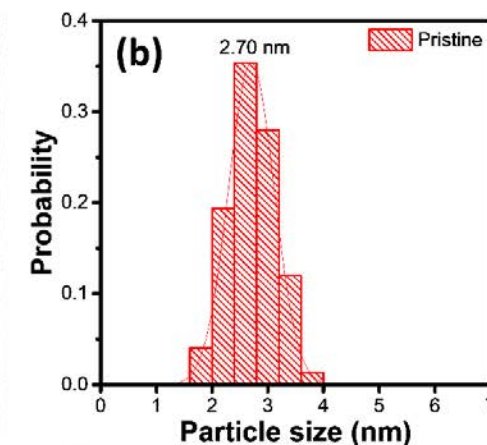
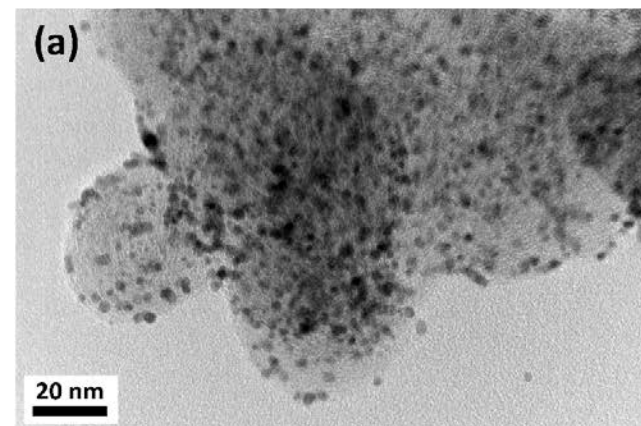
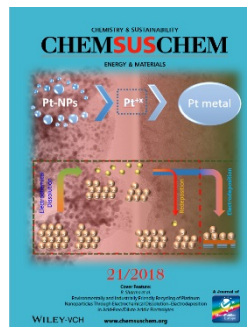
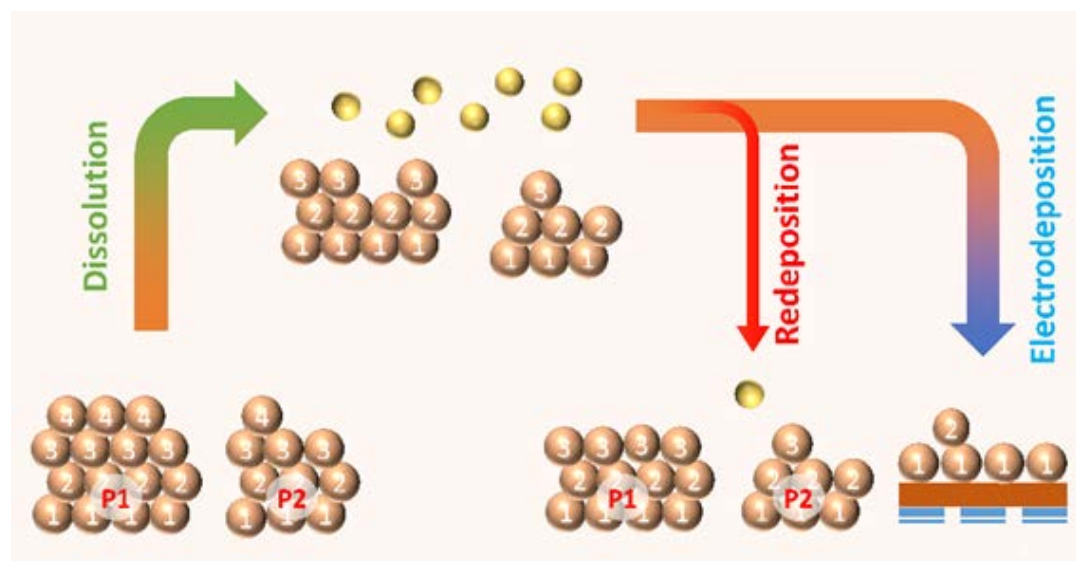
Electrochemical potential cycling enhances the dissolution rates (2-3 order) as compared to static potential

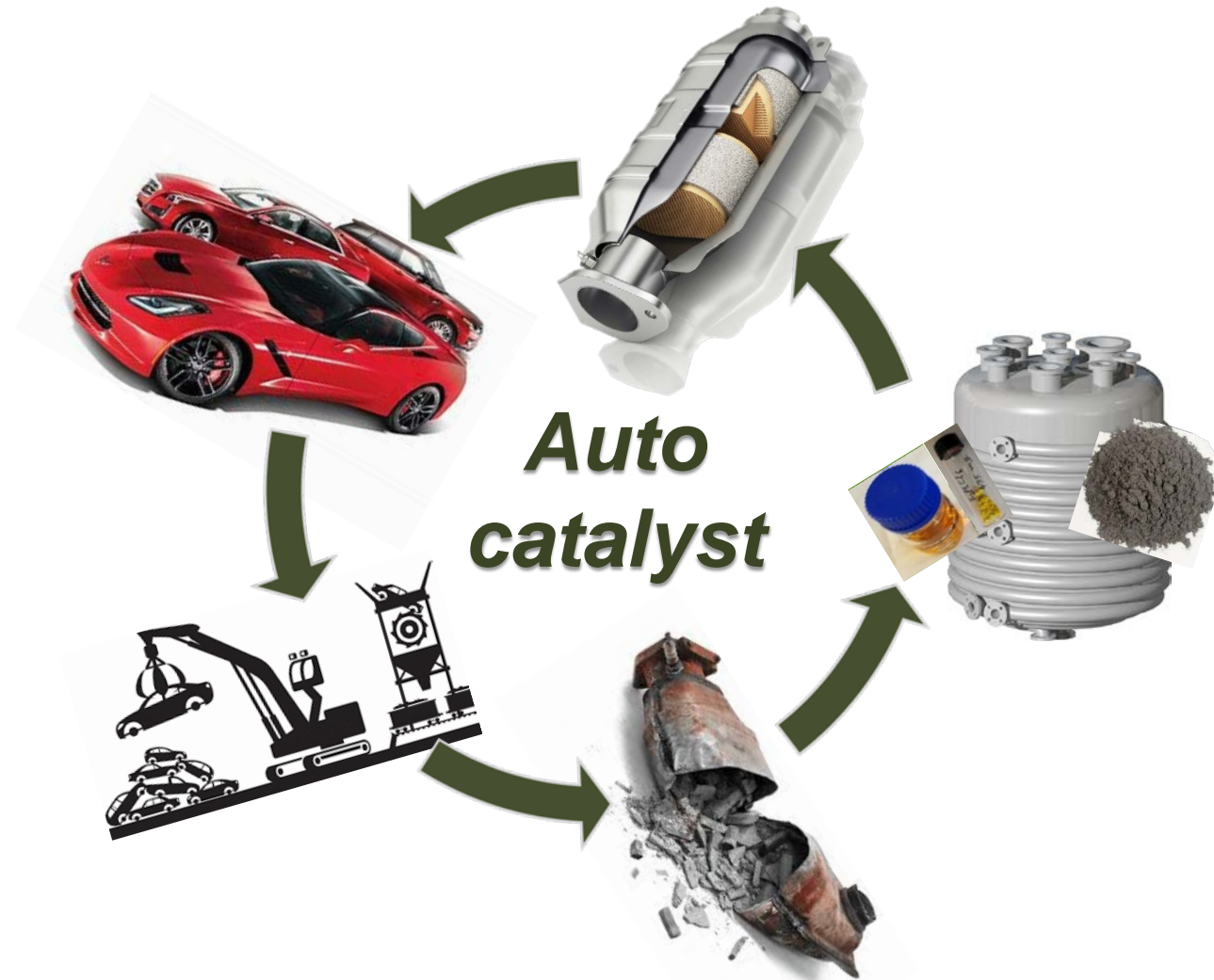
Dissolution efficiency > 95%



Ordinary potentiodynamic operation leads to significant redeposition.

Continuous collection of dissolved Pt species with a separate electrode increases dissolution efficiency and improve reaction kinetic.





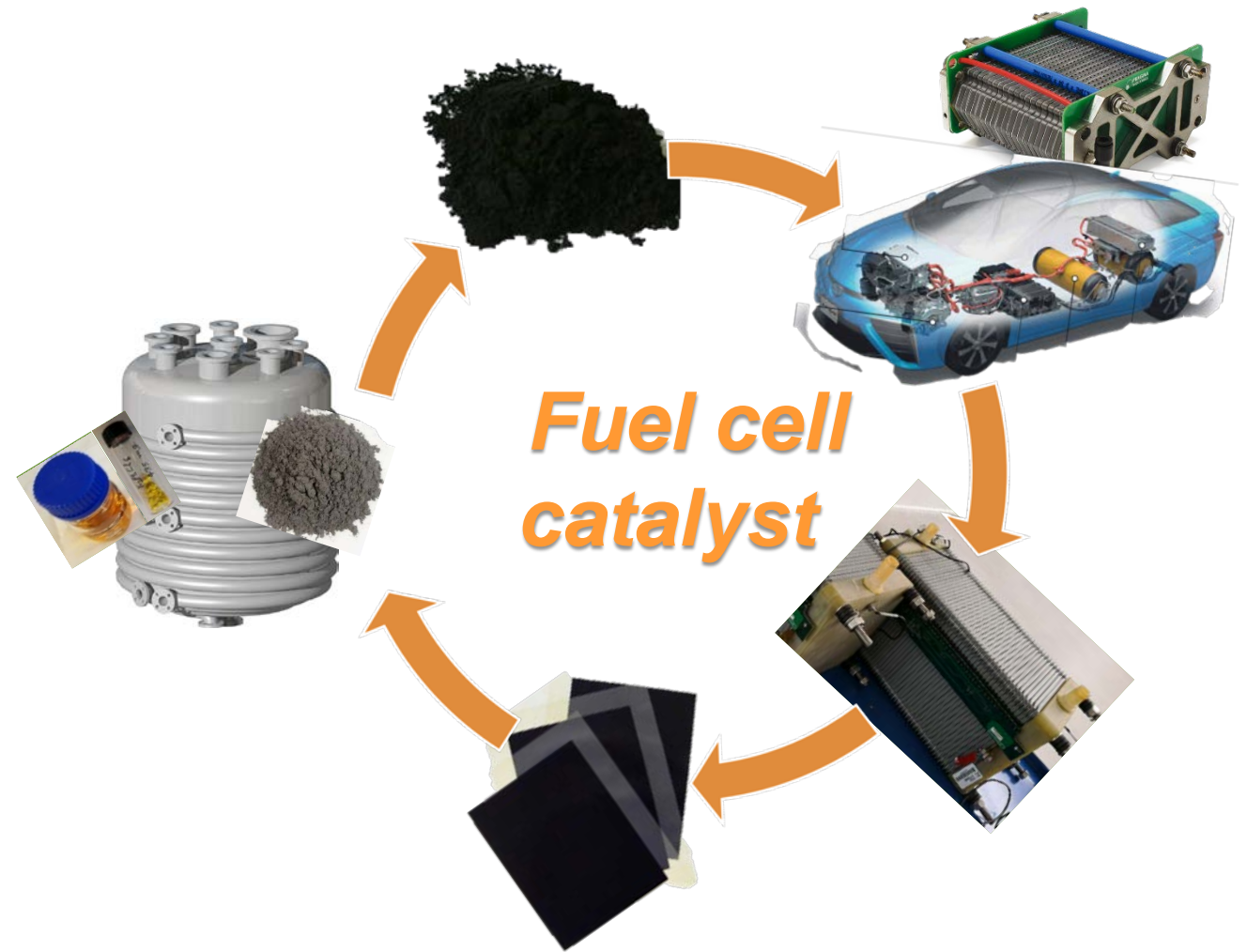
Economy and benefit

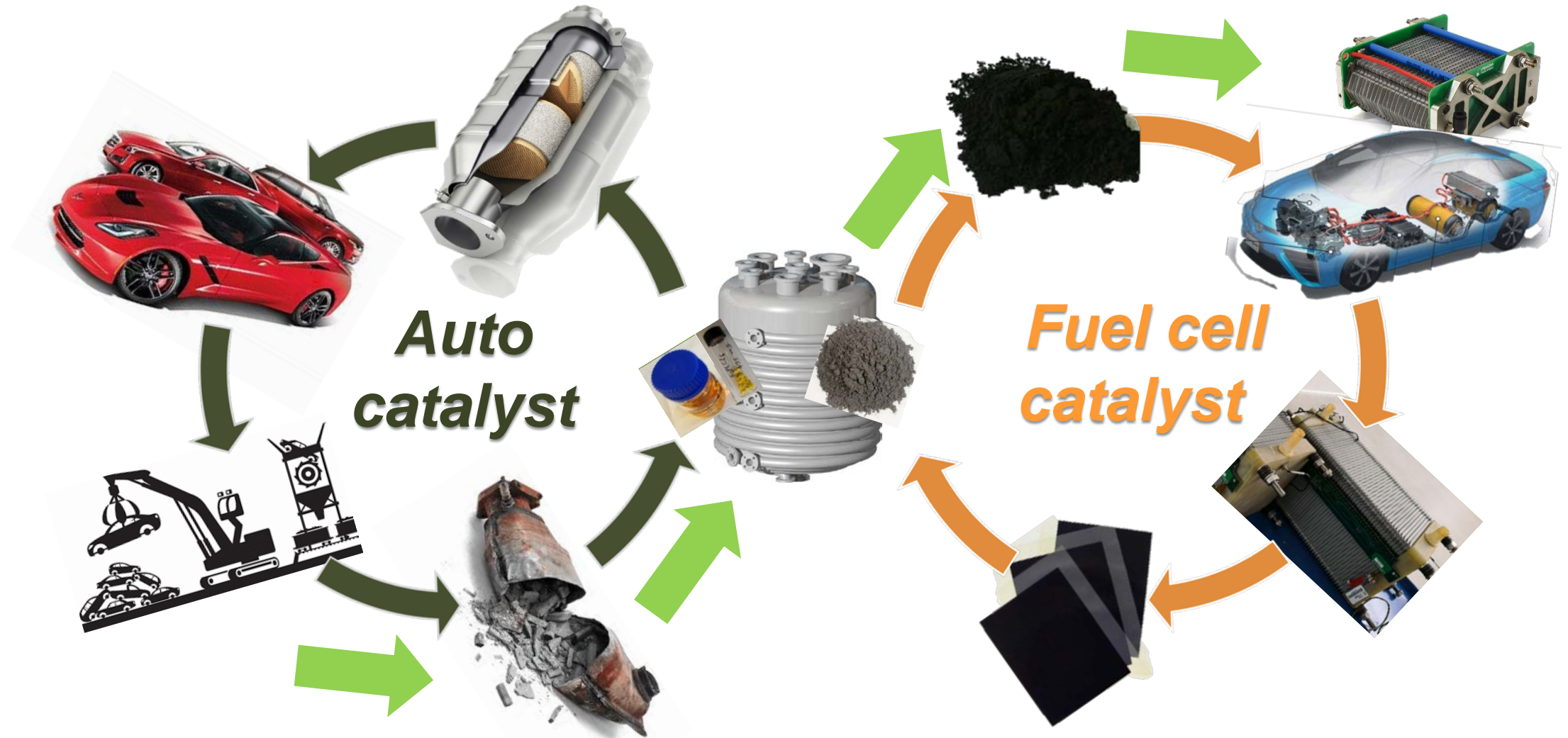
Secondary raw materials

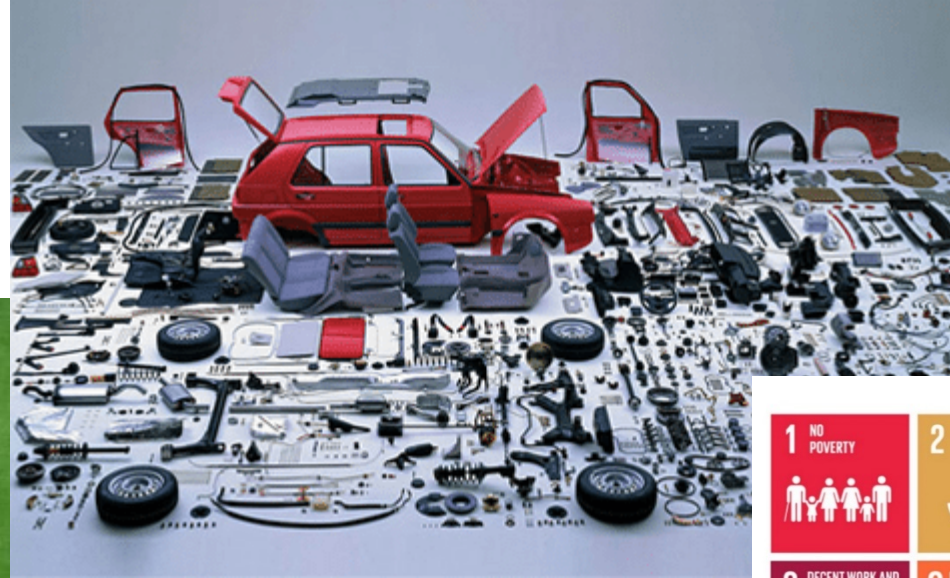
- 90% cheaper than primary production
- Reduce environmental and health issues
- Smart waste management – urban mining

Alternative process

- Flexible and low capital cost
- Environmentally friendly
- Potential for decentralized facility
- Value added products







- Improve waste management and regulation
- Sustainable recycling development
- Green transformation

Acknowledge



Thanks for your attention
mashu@kbm.sdu.dk